

Empowering Deammonification Process Controls with Direct N₂O Monitoring

The new possibility of measuring nitrous oxide (N₂O) in the deammonification process yields important insights about the anammox bacteria substrate availability. N₂O is tightly linked to the nitrite (NO₂⁻) concentration, the key substrate besides ammonium (NH₄⁺). To address the challenging control of the deammonification process, wastewater companies have invested in measurement technology from Unisense Environment. This enables them to measure N₂O levels directly in the process tanks, balance anammox bacteria substrates, and additionally document and minimize the climate impact of the deammonification process.

Background

In modern wastewater treatment plants approximately 20% of the incoming nitrogen is bound in organic form and is diverted to anaerobic digesters for biogas and energy production. Subsequently, the digester sludge is dewatered and the sludge reject water contains the diverted nitrogen in form of concentrated ammonium, typically around 1 g/L. Conventionally, the reject water has been added back to the mainstream processes. However, due to the very low COD/N ratio in the reject water this imposes a strain on the mainstream processes costing more aeration energy and in some cases carbon dosing to fulfill stringent nitrogen removal requirements in the effluent. Henceforth, separate side stream treatment processes have gained much attention and in particular the deammonification process has proven effective for side stream removal of nitrogen leading to reduction of energy for aeration and carbon needed for denitrification. The deammonification process is therefore often coupled with a reduced CO₂ footprint.

Nitrous oxide (N₂O), which is produced from an incomplete nitrogen conversion during wastewater treatment, is one of the wastewater industry's biggest environmental offenders with a carbon footprint 300 times higher than that of CO₂. Often the climate impact from N₂O surpasses the impact from the entire WWTP energy consumption. Conventionally, a key number of 0.5% of the influent nitrogen has been used to calculate the CO₂ footprint from N₂O. However, recent publications have demonstrated that the N₂O emission from side stream processes is much higher, typically in the range of 1% to +6% of the influent nitrogen. This increased N₂O production can easily deteriorate the positive effect from energy savings in the side stream processes. Furthermore, the control strategies for deammonification processes often rely on advanced analytical sensors that have short calibration intervals and high maintenance costs associated with them.

Optimization of the process

Measuring the N₂O parameter in the deammonification process yields important insights about the anammox bacteria substrate availability. **Firstly**, N₂O is tightly linked to the nitrite (NO₂⁻) concentration, the key substrate besides NH₄⁺, for the anammox bacteria. **Secondly**, the amount of oxygen available for NH₄⁺ conversion to NO₂⁻ is important for balancing the deammonification process. **Thirdly**, measuring the nitrite level is difficult with present sensors and will be indirect as nitrite is formed and consumed inside the biofilm or granulas. Unlike nitrite, N₂O is not consumed by the anammox bacteria and therefore online N₂O monitoring will be a more precise measure of

the substrate balance. In the event that too high ammonium oxidation rates lead to nitrite accumulation, N₂O will build up in the liquid as a warning. **Finally**, N₂O is clearly linked to the ammonium loading of the process and no straight correlation between N₂O and pH is obvious. Therefore, the N₂O Wastewater Sensor has proven to be much more reliable than nitrates and pH sensor data for optimizing the deammonification process controls and for N₂O emission reduction.

Deammonification process data example

A number of researchers have shown that there is a correlation between the ammonium conversion rate and the N₂O production rate. In these studies they showed that an increase in the ammonium oxidation rate led to an increased N₂O formation and derived emission due to aeration stripping. Furthermore, transient aeration sequences between low or anoxic oxygen levels to a higher aeration level seemed to be enhancing the N₂O emission rate and pointed to a more continuous aeration strategy.

In the one week dataset below (10 second data interval) the sequence of aeration was gradually changed towards a lower but steady aeration. Monitoring N₂O (**Figure 1, red curve**) noticeably demonstrates that the concentration and emission gradually diminishes to less than 10% of the initial N₂O level. Also, the nitrite level was below 5 mg-N/mL throughout the entire period, and was not the primary cause of the large N₂O concentration and emission.

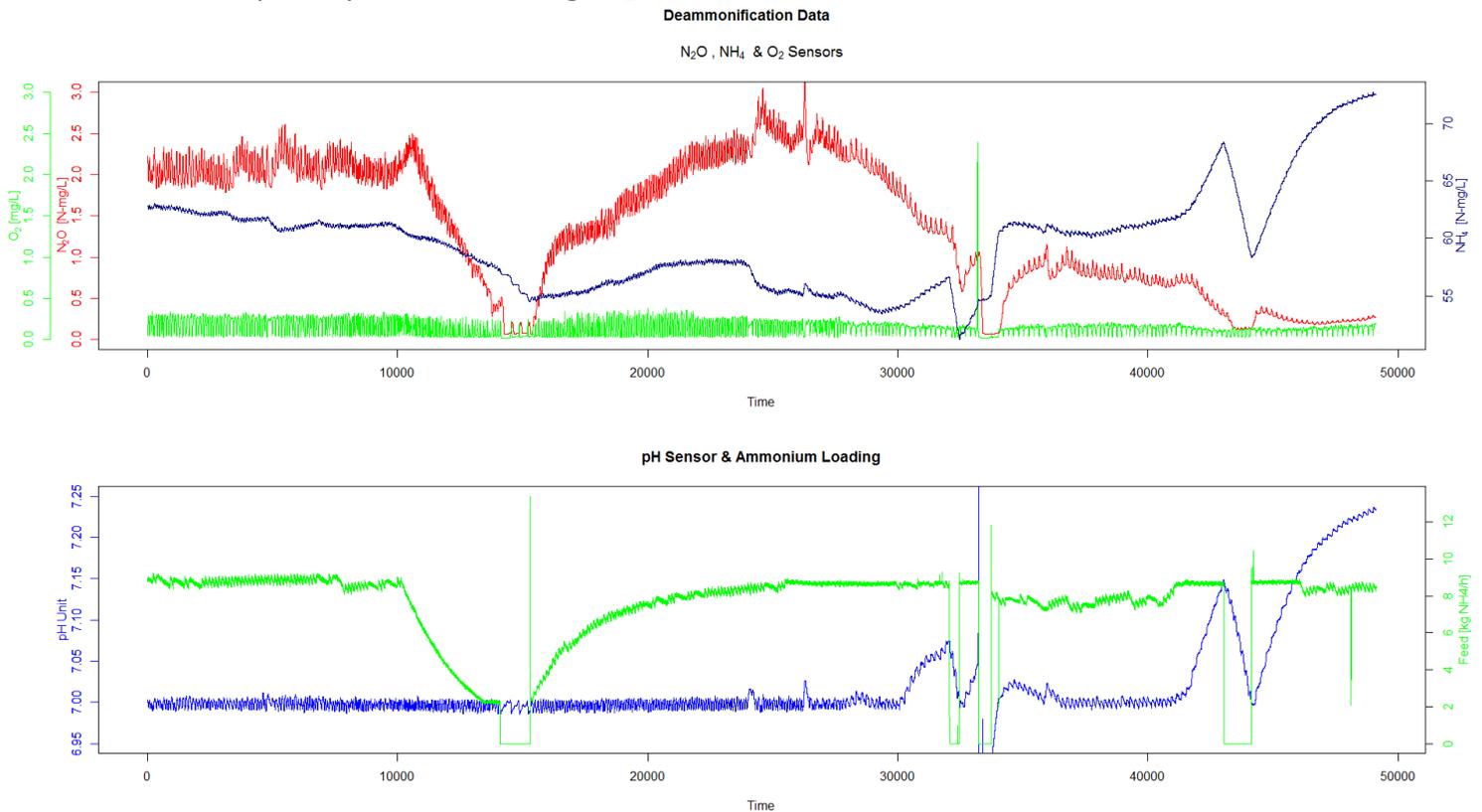


Figure 1: Example of deammonification monitoring for one week with the N₂O Wastewater Sensor (red), O₂ (green) and relation to other parameters such as NH₄⁺ (dark blue), pH (light blue). Monitoring was done through a trial run where different aeration and feed rates, and pH values were tested. N₂O in the process tank closely followed the loading of ammonium (panel B; feed rate in green) and was not clearly correlated to pH.

When pH is allowed to migrate upwards it is generally a benefit for the bacteria as the maximum substrate conversion activity increases for both the ammonium oxidizing bacteria and the anammox bacteria. In **figure 1**, the pH was allowed to increase at the end of the 1 week period and although the ammonium concentration increased due to the lower aeration discussed above, it levels out. The ammonium loading of the system was the same as in the beginning of the week. Therefore, the same substrate oxidation rate is observed. But as the ammonium oxidizing bacteria are not at their maximum oxidation rate at the new pH, much less production of N₂O is observed. Finally, a lower ammonium concentration in the process was tested through a lower continuous federate that kept the process substrate limited resulting in lower N₂O, seen 1/3rd into the week.

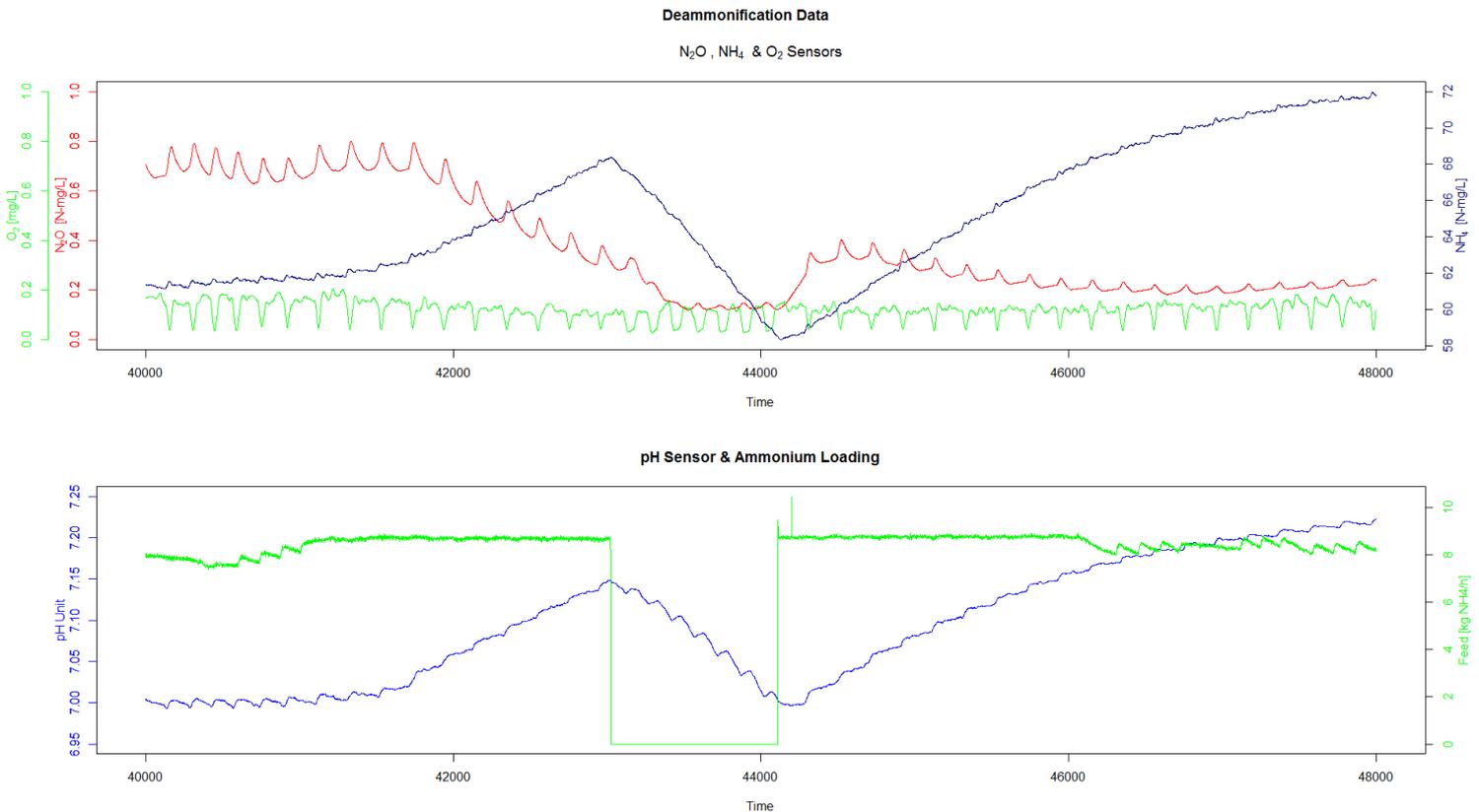


Figure 2: Example of a loading control based on the N₂O signal (red) with floating pH value (light blue) and disabled pH control schema. NH₄⁺ (dark blue), O₂ (green), and feed rate (Panel B, green). The aeration is almost continual in this particular test.

In **figure 2**, an enlargement of the last 2 days of the monitoring period is depicted demonstrating the transition from a slightly higher and sequential aeration to a lower and ultimately steady aeration. The oxygen change is small but the effect on the N₂O level is clear. Controlling the deammonification process by balancing the ammonium and nitrite substrates through inclusion of N₂O in the control scheme can therefore decrease the N₂O emission many fold.

Control strategies possible with online input from direct N₂O monitoring

- **Implement continuous influent and aeration strategies**
 - Steady state inflow and aeration.
 - No perturbation from aeration pauses and no lag phases.
 - Load control based on N₂O sensor and steady-state N₂O sensor signal.
 - Aeration control based on N₂O sensor.
- **Implement multi parameter aeration control**
 - Too high NH₄⁺ turnover leads to NH₄⁺/NO₂⁻ imbalance.
 - High NO₂⁻ leads to high N₂O and N₂O-emission
 - N₂O works as ***proxy sensor for NO₂⁻***
- **Unstable and drifting pH control**
 - pH uncertainties lead to AOB and anammox inhibition.
 - AOB and anammox activity drops dramatically at pH below 7.0.
- **Avoid NO₂⁻ accumulation and inhibition**
 - > 5 mg-N/L nitrite leads to slow 'irreversible' inhibition of the anammox biomass.
 - ***Can happen unnoticed*** as substrate limited biomass deeper in the granules initially takes over from newly inhibited outer biomass eventually leading to sudden process breakdown, when the whole granule has been inhibited.
 - Keep NO₂⁻ accumulation in check through reliable N₂O monitoring.
 - Works at any biomass concentration and without any specific calibration.
 - Keep heterotrophs in check.
 - Higher COD/N ratios will lead to heterotrophic growth and incomplete denitrification resulting in high N₂O.
- **Consider recycling the aeration gas**
 - Better mixing, biofilm and granola size control without over-aeration.
 - Lower N₂O and better O₂ control.
 - Control air replenishment based on NH₄, N₂O, and pH rates of change.

Aim for a low steady state N₂O level and act on high N₂O